

# How Cells Harvest Chemical Energy

# 6

You need to eat and breathe because your cells need food and oxygen for energy. In every cell in your body, organic molecules and oxygen interact in a complex process called cellular respiration. In this process, food molecules such as glucose are broken down and the energy contained in their chemical bonds is used to make ATP. The ATP made in cellular respiration is then used to drive cellular activities. Right now, ATP produced in cellular respiration is being used to generate the nerve impulses from your eyes to your brain, to move your muscles, and to drive your heartbeat. This chapter explains how your cells harvest the energy that keeps you alive.

## Organizing Your Knowledge

### Exercise 1 (Introduction – Module 6.3)

Review the basic terms and concepts of cellular respiration by filling in the blanks below.

Right now, you are breathing at a steady rate of 12 to 20 breaths per minute. Breathing, or <sup>1</sup> \_\_\_\_\_, is necessary for life, but why? Breathing allows the body to take in <sup>2</sup> \_\_\_\_\_ gas and expel waste <sup>3</sup> \_\_\_\_\_. Your breathing is closely related to <sup>4</sup> \_\_\_\_\_, the aerobic harvest of the energy in food molecules by cells. Most of the time, most cells acquire energy by taking in both <sup>5</sup> \_\_\_\_\_ and <sup>6</sup> \_\_\_\_\_ from the blood. These two substances interact, the sugar is broken apart, and <sup>7</sup> \_\_\_\_\_ and <sup>8</sup> \_\_\_\_\_ are produced. In the process, some of the energy is stored in molecules of \_\_\_\_\_, which provide the energy for body activities. To make enough ATP for their needs, average human beings must take in food that provides about \_\_\_\_\_ kilocalories (kcal) of energy per day.

When using oxygen, cells are said to function <sup>11</sup> \_\_\_\_\_. Cells making ATP this way capture about <sup>12</sup> \_\_\_\_\_ of the energy in glucose. For short bursts, cells can make ATP <sup>13</sup> \_\_\_\_\_ — that is, without using oxygen. This process is much less efficient; it only banks about <sup>14</sup> \_\_\_\_\_ % of the energy in glucose. But it is useful during intense bursts of activity, such as sprinting or heavy lifting.

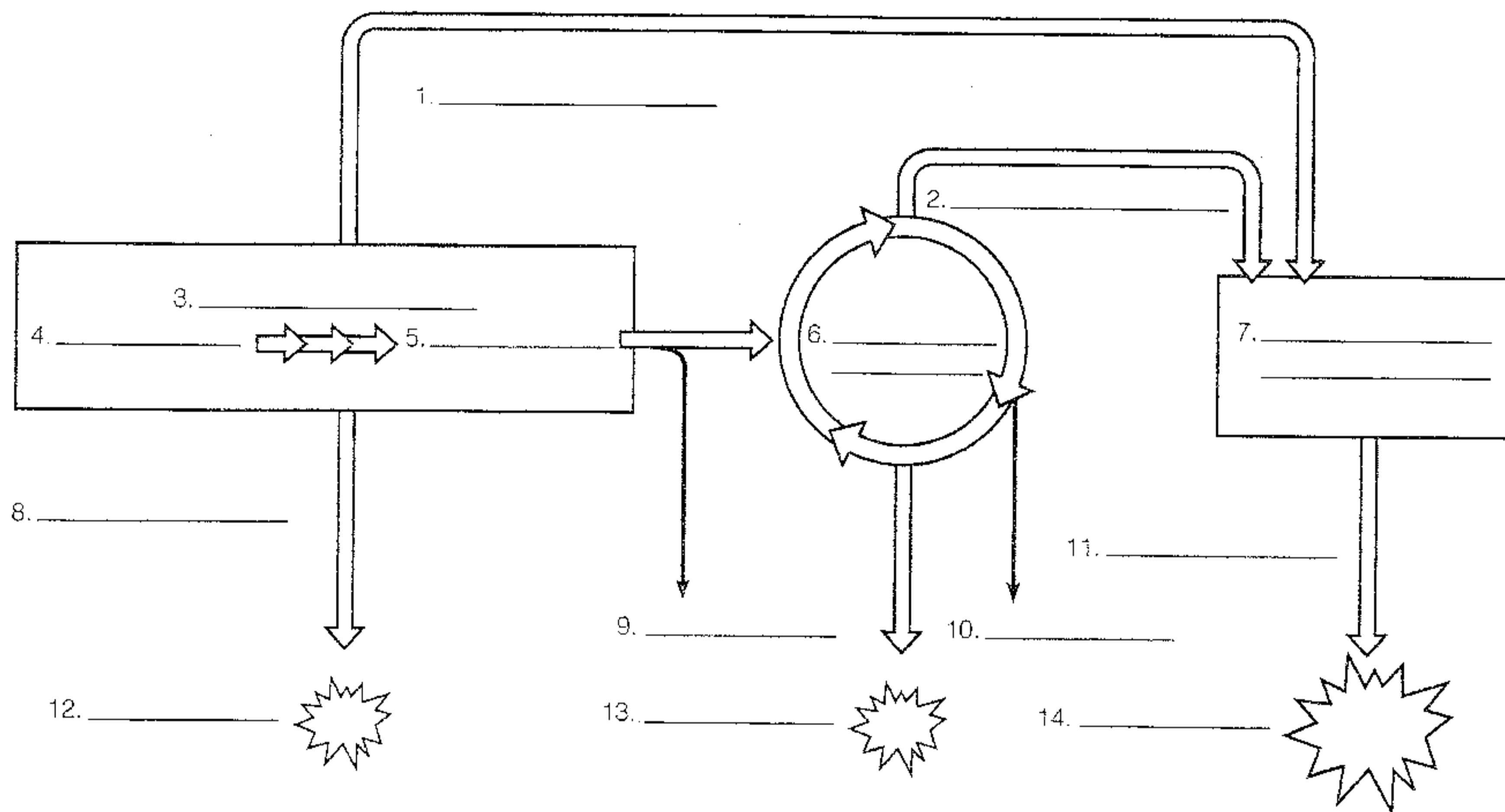
Muscles contain a mixture of two kinds of cells, or fibers, specialized either for aerobic or anaerobic ATP production. <sup>15</sup> \_\_\_\_\_ fibers can sustain repeated, long contractions, by continuously producing ATP via <sup>16</sup> \_\_\_\_\_ cellular respiration. Slow fibers are long and <sup>17</sup> \_\_\_\_\_, maximizing their surface area and contact with nearby <sup>18</sup> \_\_\_\_\_ that deliver oxygen. They have many \_\_\_\_\_, the structures where aerobic ATP breakdown occurs. And they \_\_\_\_\_ in <sup>20</sup> \_\_\_\_\_, a red protein related to hemoglobin that supplies O<sub>2</sub>

molecules. The <sup>21</sup> \_\_\_\_\_ meat of a turkey leg consists mostly of myoglobin-rich slow muscle fibers. The white meat of a turkey breast, on the other hand, consists mostly of <sup>22</sup> \_\_\_\_\_ muscle fibers, which are specialized for quick, powerful bursts of flight. These fibers are <sup>23</sup> \_\_\_\_\_ and are more powerful than slow fibers, having <sup>24</sup> \_\_\_\_\_ mitochondria and <sup>25</sup> \_\_\_\_\_ myoglobin than slow fibers. During intense activity, when the blood cannot deliver O<sub>2</sub> fast enough for aerobic cellular respiration, fast fibers can function anaerobically, making small amounts of ATP without oxygen. They don't completely break down <sup>26</sup> \_\_\_\_\_ and therefore do not capture all its energy, and instead of producing CO<sub>2</sub> they make <sup>27</sup> \_\_\_\_\_, which makes muscles <sup>28</sup> \_\_\_\_\_ and fatigue. This is why <sup>29</sup> \_\_\_\_\_ fibers are best at producing short bursts of power.

Human muscles contain both kinds of fibers. Their proportions vary from muscle to muscle, and person to person. A runner whose leg muscles are primarily composed of <sup>30</sup> \_\_\_\_\_ fibers would be more likely to excel in distance events, while an individual with an abundance of <sup>31</sup> \_\_\_\_\_ fibers might make a better sprinter.

**Exercise 3 (Module 6.8)**Web/CD Activity 6A *Overview of Cellular Respiration*

The illustration in Module 6.8 introduces the three stages of cellular respiration. After studying it, see if you can label the diagram below without referring to the text. Include **electron transport chain and chemiosmosis, pyruvic acid, mitochondrion, CO<sub>2</sub>, high-energy electrons carried by NADH, Krebs cycle, glycolysis, cytoplasmic fluid, ATP, glucose, and NADH and FADH<sub>2</sub>**. (Note: 3, 6, and 7 are processes, 8 and 11 are places, and the rest are inputs and outputs.)

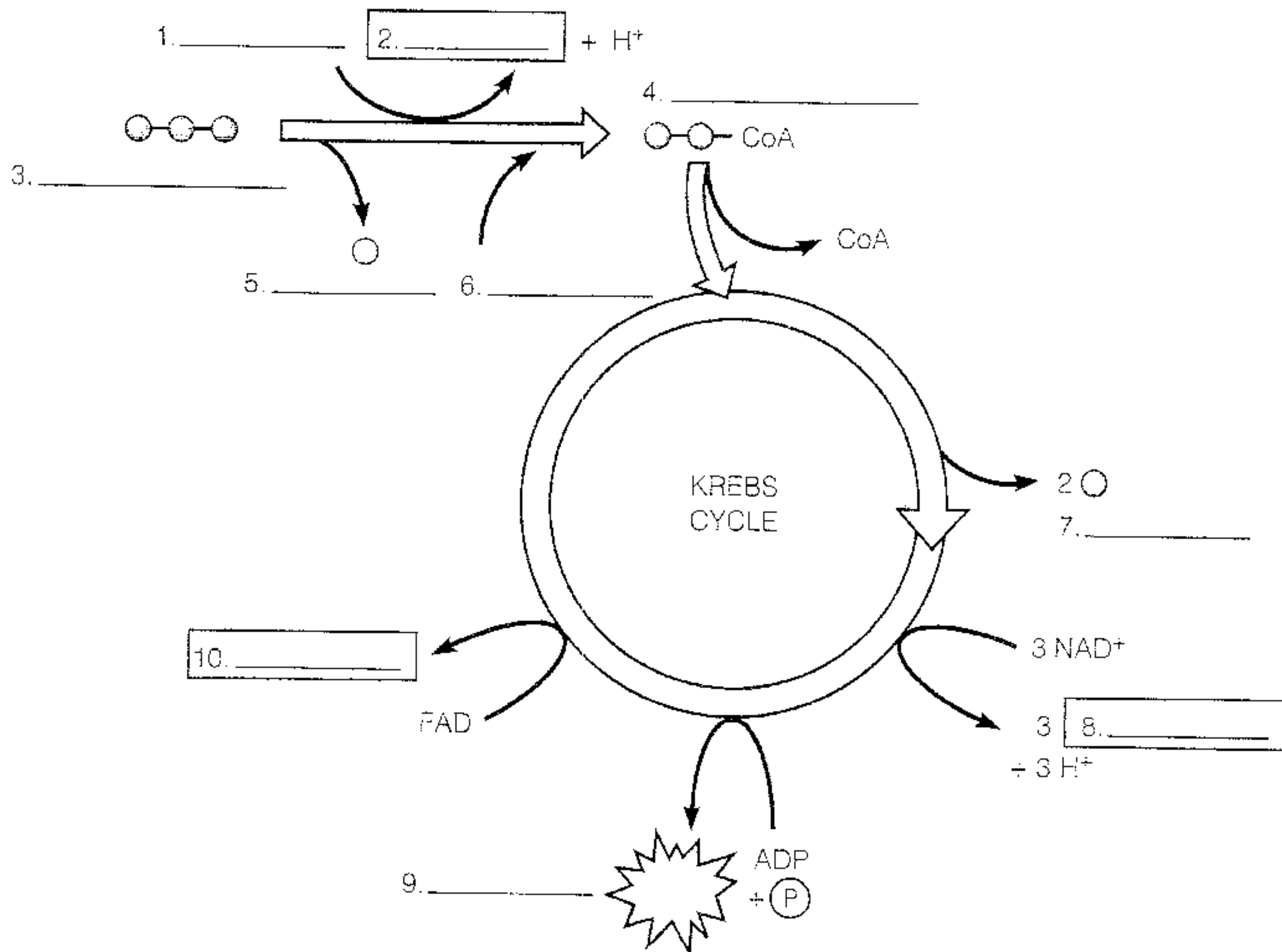
**Exercise 4 (Module 6.9)**Web/CD Activity 6B *Glycolysis*

Glycolysis is the first of three steps in cellular respiration. Review glycolysis by matching each phrase on the right with a term on the left. Some terms are used twice.

- |                     |       |  |
|---------------------|-------|--|
| A. NADH             | _____ | 1. Compound formed between glucose and pyruvic acid          |
| B. Pyruvic acid     | _____ | 2. Not involved in glycolysis                                |
| C. ATP              | _____ | 3. Fuel molecule broken down in glycolysis                   |
| D. NAD <sup>+</sup> | _____ | 4. Produced by substrate-level phosphorylation               |
| E. Glucose          | _____ | 5. Invested to energize glucose molecule at start of process |
| F. Glycolysis       | _____ | 6. Reduced as glucose is oxidized                            |
| G. ADP and P        | _____ | 7. Glucose converted to two molecules of this                |
| H. Oxygen           | _____ | 8. Assembled to make ATP                                     |
| I. Intermediate     | _____ | 9. "Splitting of sugar"                                      |
|                     | _____ | 10. Carries hydrogen and electrons from oxidation of glucose |

**Exercise 5 (Modules 6.10 – 6.11)**Web/CD Activity 6C *The Krebs Cycle*

Pyruvic acid from glycolysis is chemically altered and then enters the Krebs cycle, a series of steps that completes the oxidation of glucose. The energy of pyruvic acid is stored in NADH and FADH<sub>2</sub>. To review these processes, fill in the blanks in the diagram below. (Try to do as many as you can without referring to the text.) Include the following: NAD<sup>+</sup>, pyruvic acid, CO<sub>2</sub>, FADH<sub>2</sub>, NADH, coenzyme A, ATP, and acetyl CoA.

**Exercise 6 (Module 6.12)**Web/CD Activity 6D *Electron Transport and Chemiosmosis*

Circle the correct words or phrases in parentheses to complete each sentence.

The <sup>1</sup> (*final, second*) stage of cellular respiration is the electron transport chain and synthesis of <sup>2</sup> (*glucose, ATP*) by <sup>3</sup> (*chemiosmosis, active transport*). The electron transport chain is a sequence of <sup>4</sup> (*electron, proton*) carriers built into the <sup>5</sup> (*outer, inner*) membrane of the mitochondrion. Molecules of <sup>6</sup> (*ADP, NADH*) bring hydrogens and electrons to the chain from glycolysis and <sup>7</sup> (*the Krebs cycle, chemiosmosis*). The electrons move along the chain from carrier to carrier in a series of redox reactions, finally joining with <sup>8</sup> (*H<sub>2</sub>O, CO<sub>2</sub>, O<sub>2</sub>*) and H<sup>+</sup> from the surrounding solution to form <sup>9</sup> (*H<sub>2</sub>O, CO<sub>2</sub>, O<sub>2</sub>*). Energy released by the electrons is used to move protons—<sup>10</sup> (*H<sup>+</sup> ions, ADP molecules*)—by <sup>11</sup> (*active transport, passive transport*) into the space between the inner and outer mitochondrial membranes.

The buildup of protons in the intermembrane space—a proton gradient—constitutes <sup>12</sup> (*kinetic, potential*) energy that the cell can tap to make <sup>13</sup> (*ATP, glucose*). The concentration of protons tends to drive them back through the membrane into the <sup>14</sup> (*inner*

compartment of the mitochondrion, cytoplasm of the cell), but protons can cross the membrane only by passing through special protein complexes, called <sup>15</sup> (coenzyme As, ATP synthases). As these complexes allow protons back through the membrane, their enzymes harness the energy of the moving protons to phosphorylate <sup>16</sup> (NAD, ADP) and make <sup>17</sup> (NADH, ATP). Thus, electron transport and chemiosmosis transform <sup>18</sup> (most, some, a small portion) of the energy extracted from glucose into the phosphate bonds of ATP.

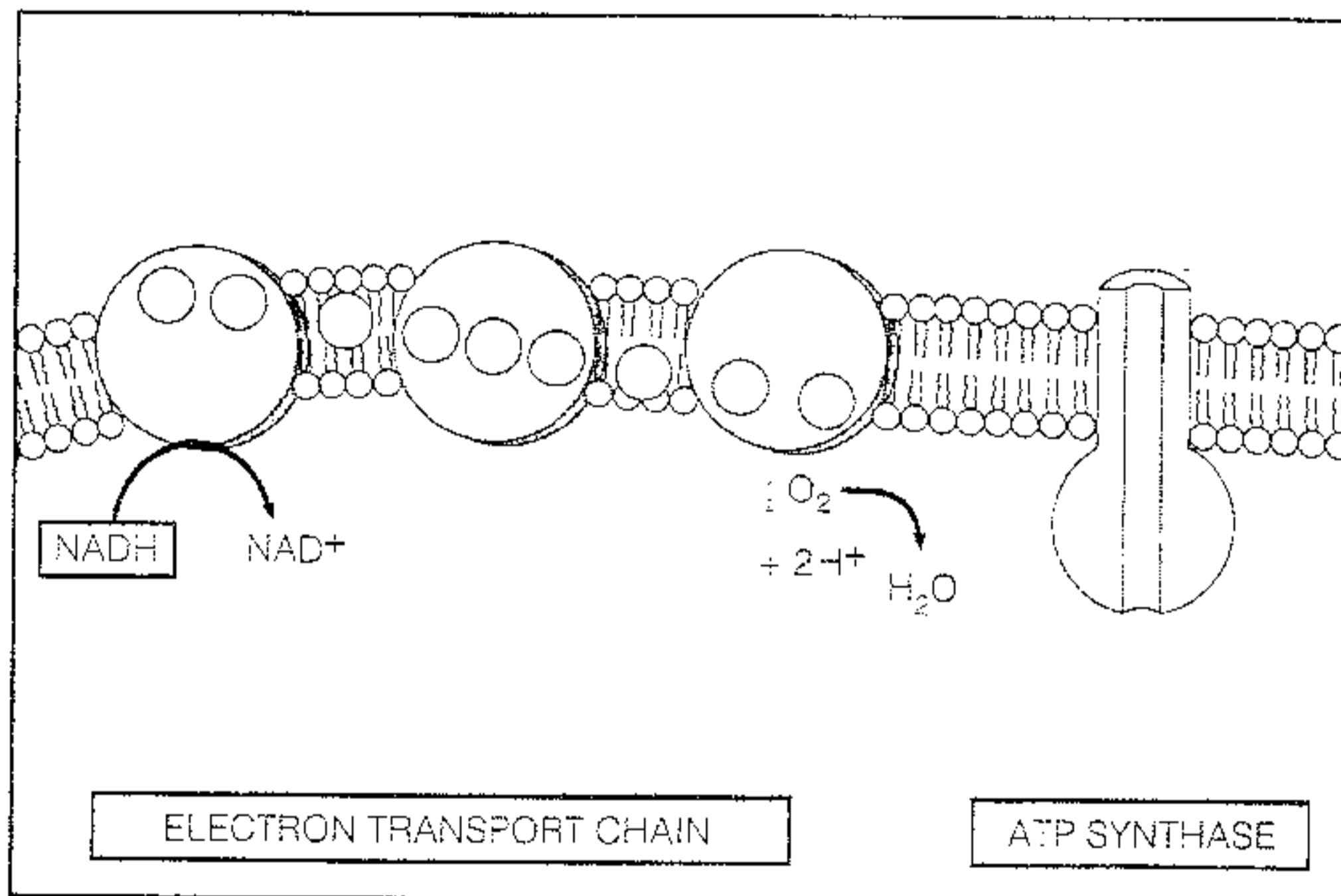
### Exercise 7 (Modules 6.12 - 6.13)

#### Web/CD Activity 6D Electron Transport and Chemiosmosis

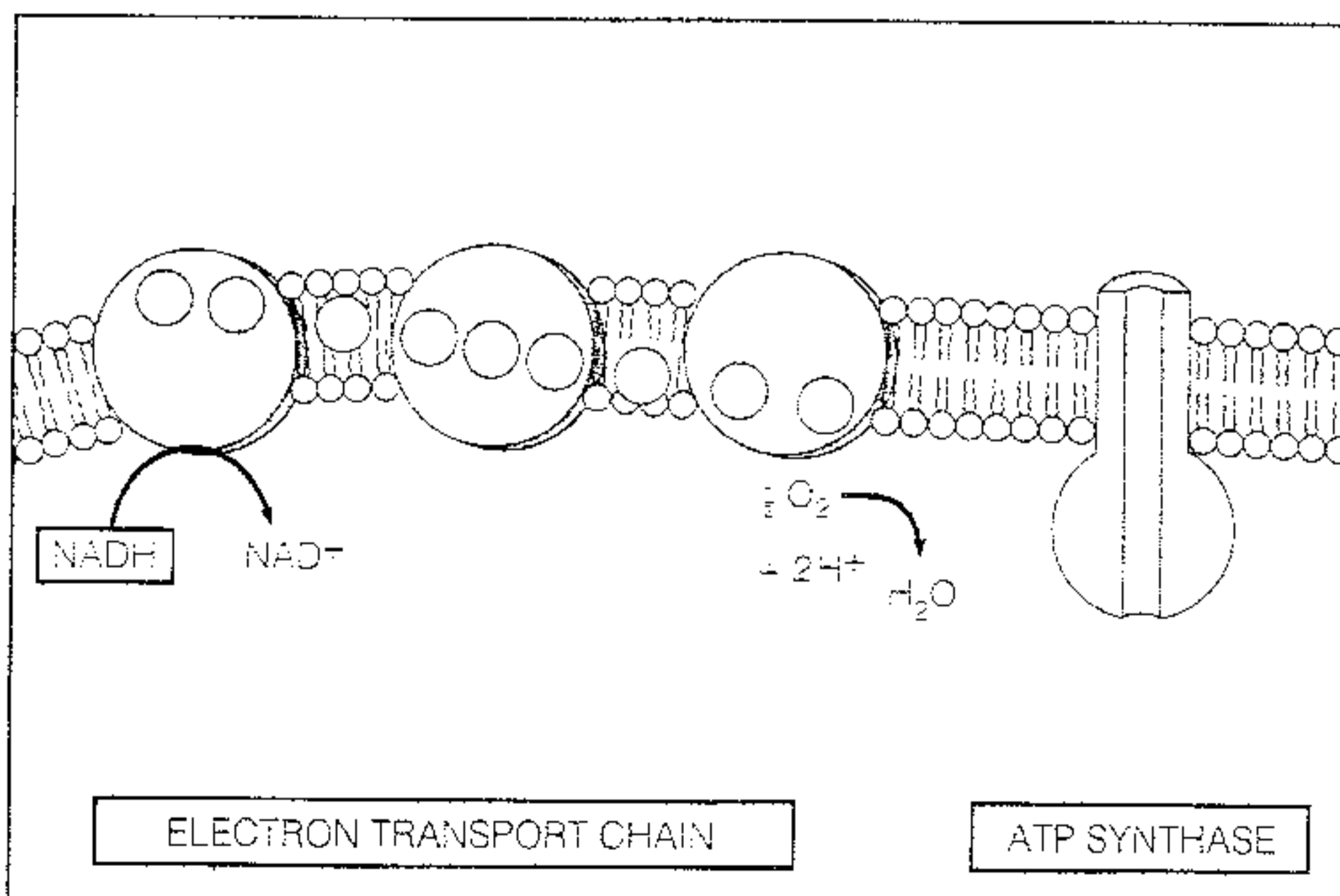
These diagrams will help you review electron transport, chemiosmosis, and how poisons disrupt them. In the first diagram, show how the processes work normally. Trace movement of an electron with an orange arrow, movement of  $H^+$  ions (active transport and chemiosmosis) with black arrows, and formation of ATP with a pink arrow.

In the second diagram, draw arrows showing the movement of electrons and  $H^+$  and the formation of ATP, as in the first diagram. Then draw a red line where each poison acts, to show how each of the poisons short-circuits the normal processes. Label the poisons rotenone, cyanide, carbon monoxide, DNP, and oligomycin.

#### 1. Normal electron transport and chemiosmosis:



#### 2. Effects of poisons:



**Exercise 8 (Modules 6.8 – 6.14)**

- Web/CD Activity 6A *Overview of Cellular Respiration*  
 Web/CD Activity 6B *Glycolysis*  
 Web/CD Activity 6C *The Krebs Cycle*  
 Web/CD Activity 6D *Electron Transport and Chemiosmosis*

Check your overall understanding of cellular respiration by matching each of the phrases below with one of the three stages of the process. Use G for glycolysis, K for Krebs cycle, and E for electron transport and chemiosmosis.

- \_\_\_\_\_ 1. Generates most of the ATP formed by cellular respiration  
 \_\_\_\_\_ 2. Begins the oxidation of glucose  
 \_\_\_\_\_ 3. Occurs outside the mitochondrion  
 \_\_\_\_\_ 4. Produces 4 ATPs per glucose by substrate-level phosphorylation, but 2 ATPs per glucose are used to get it started  
 \_\_\_\_\_ 5. Oxidizes NADH and FADH<sub>2</sub>, producing NAD<sup>+</sup> and FAD  
 \_\_\_\_\_ 6. Carried out by enzymes in the matrix (fluid) of the mitochondrion  
 \_\_\_\_\_ 7. Here electrons and hydrogen combine with O<sub>2</sub> to form H<sub>2</sub>O  
 \_\_\_\_\_ 8. Occurs along the inner mitochondrial membrane  
 \_\_\_\_\_ 9. Generates most of the CO<sub>2</sub> produced by cellular respiration  
 \_\_\_\_\_ 10. FADH<sub>2</sub> and NADH deliver hydrogen ions and electrons to this stage  
 \_\_\_\_\_ 11. ATP synthase makes ATP  
 \_\_\_\_\_ 12. Reduces NAD<sup>+</sup> and FAD, producing NADH and FADH<sub>2</sub>

**Exercise 9 (Module 6.15)**

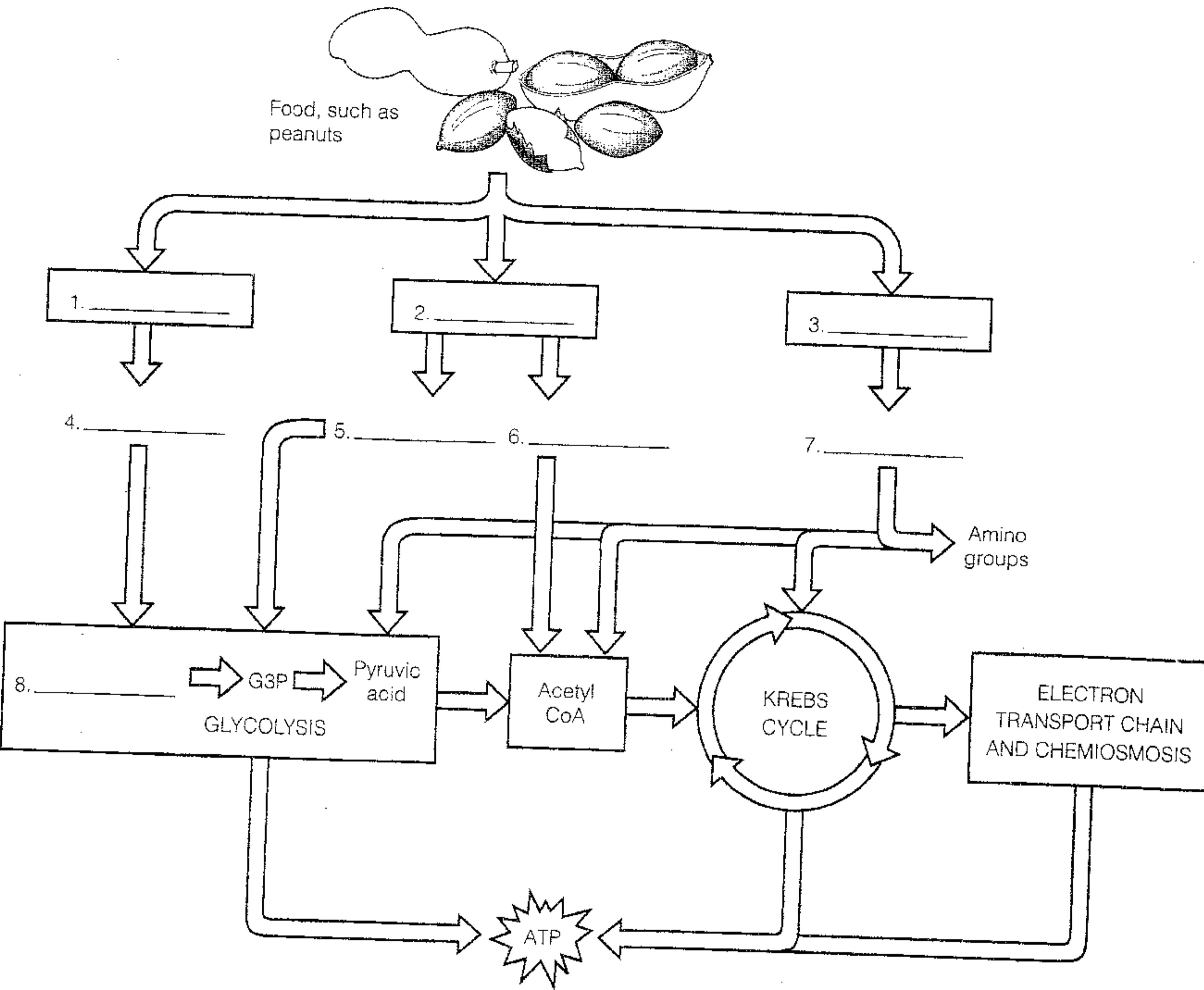
- Web/CD Activity 6E *Fermentation*

Review fermentation by filling in the blanks below.

- \_\_\_\_\_ anaerobes can make their ATP by fermentation or aerobic respiration.
- \_\_\_\_\_ is an organism that normally uses aerobic respiration to produce ATP, but it can generate ATP without oxygen, via alcoholic fermentation.
- Fermenters replenish their supply of NAD<sup>+</sup> by using NADH to oxidize \_\_\_\_\_ acid.
- When oxygen is scarce, human \_\_\_\_\_ cells can make ATP by lactic acid fermentation.
- Fermentation enables cells to make ATP in the absence of \_\_\_\_\_.
- For every molecule of glucose consumed, glycolysis produces two molecules of pyruvic acid, two molecules of ATP, and two molecules of \_\_\_\_\_.
- The waste products of alcoholic fermentation are \_\_\_\_\_ and carbon dioxide.
- \_\_\_\_\_ acid fermentation is used to make cheese and yogurt.
- Fermentation generates two \_\_\_\_\_ molecules for every molecule of glucose consumed.
- A cell can use \_\_\_\_\_ to generate a small amount of ATP, but it must somehow recycle its supply of NAD<sup>+</sup>.
- Like aerobic respiration, alcoholic fermentation produces \_\_\_\_\_ gas as a waste product.
- Strict \_\_\_\_\_ require anaerobic conditions and are poisoned by oxygen.

**Exercise 10 (Module 6.16)**

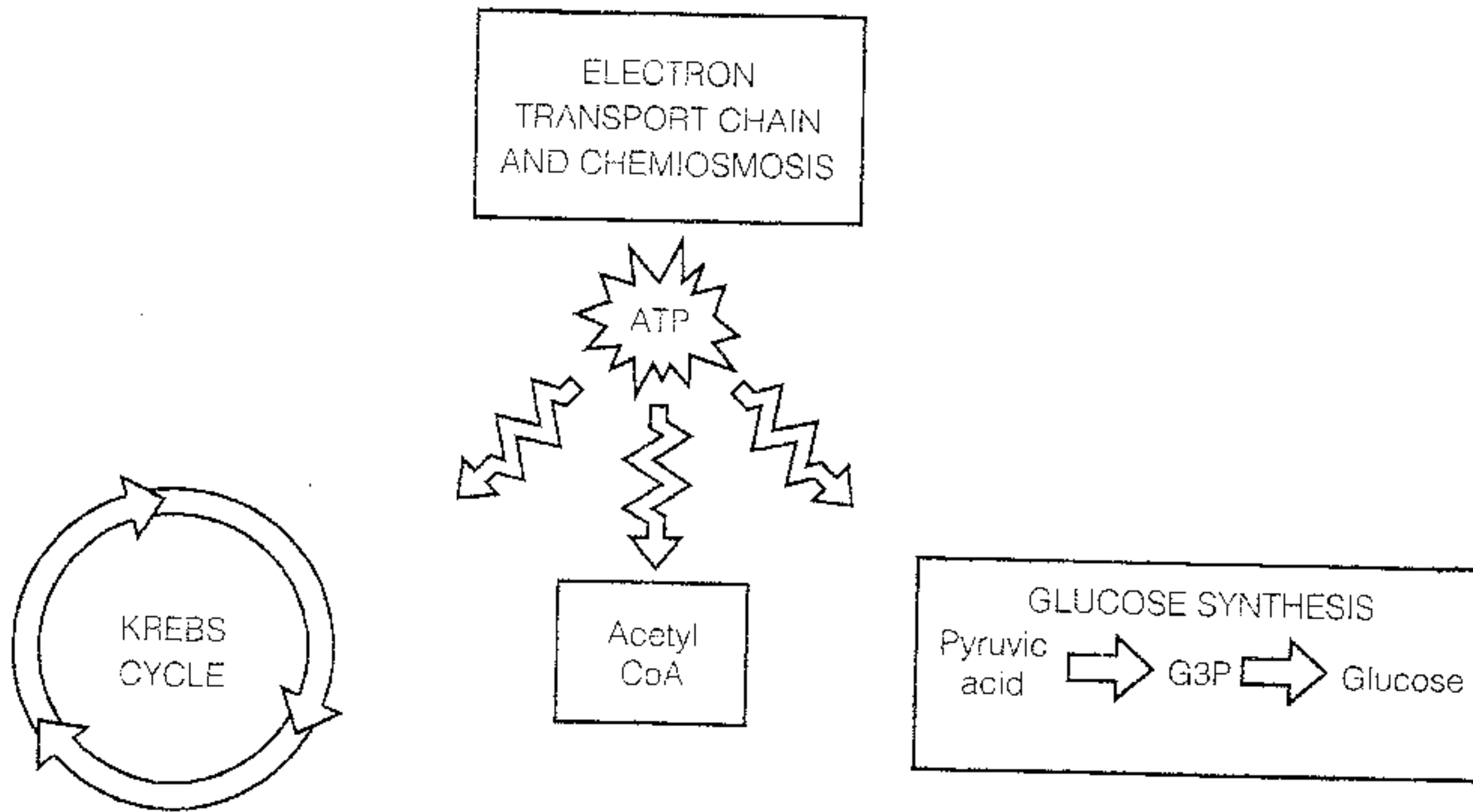
Review the molecules that can be used as fuel for cellular respiration by writing their names in the blanks in this diagram. Include **glucose, amino acids, fats, fatty acids, proteins, sugars, polysaccharides, and glycerol.**



1. Glucose  
 2. Fats, fatty acids  
 3. Amino acids  
 4. Glucose  
 5. Glycerol  
 6. Fatty acids  
 7. Amino acids  
 8. Glucose

**Exercise 11 (Module 6.17)**

Show how a cell obtains organic molecules for biosynthesis of proteins, polysaccharides, and fats by drawing the missing arrows on this diagram.



Amino groups

Amino acids

Fatty acids

Glycerol

Sugars

