The Working Cell

You can read these words because enzymes and membranes enable your cells to use energy. The light energy that bounces off the page enters your eyes and stimulates pigments held in special cell membranes. Enzymes make these pigments and convert them to a form that can absorb light. The eye cells can transmit signals through nerve cells to the brain because the membranes of these cells can selectively absorb and pump out charged particles. The energy for moving these particles comes from processes that make ATP. These processes take place through the action of enzymes on and between cell membranes. Every biological activity—not just reading, but walking, laughing, and thinking—depends on energy produced by processes that involve enzymes and membranes. Energy, enzymes, and membranes are the subjects discussed in this chapter.

Organizing Your Knowledge

Exercise 1 (Modules 5.1 - 5.5)

Web/CD Activity 5A Energy Transformations
Web/CD Activity 5B Chemical Reactions and ATP
Web/CD Activity 5C The Structure of ATP

After reading Modules 5.1–5.5, review energy, chemical reactions, and the function of enzymes by filling in the blanks in the following story.

If you were to stop eating, you would probably starve to death in weeks or months. If you were unable to breathe, you would die in minutes. Organisms need the energy that is released when food and oxygen combine. This energy is used not only to move the body but also to keep it from falling apart.

	8 apart,	
Energy is the ability to the energy that sustains living ergy of movement that is actual are able to use the energy of sulaws of ³	things. Sunlight is pure ² _ally doing work. In the proculing to produce food molecular the principles that govern erel. According to the ⁴ or transferred,	ecules. This process obeys the nergy transformations. Plants law of thermody- but it cannot be created and
No energy change is 100 are no exception to this rule. Som rather is converted to ⁷	% efficient, and the changes	that occur in photosynthesis

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⁸ law of thermodynamics says that energy changes are always accompanied	
by an increase in ⁹ , a measure of disorder. One of the reasons living	
things need a constant supply of energy is to counter this natural tendency toward disorder.	
The products of photosynthesis contain ¹⁰ potential energy	
than the reactants. This means that, overall, photosynthesis is an ¹¹	
reaction. Such a reaction consumes energy, which in photosynthesis is supplied by the	
sun.	
Photosynthesis produces food molecules, such as glucose, which store energy. Ar	1
animal might obtain this food by eating a plant or an animal that has eaten a plant. The fo	od
molecules enter the animal's cells, where their potential energy is released in the process of	of.
cellular respiration. The products of this chemical reaction (actually a series of reactions)	
contain less potential energy than the reactants. Therefore, cellular respiration is an	
process; it ¹³ energy. In fact, this is the same overal	1
change that occurs when glucose in a piece of wood or paper burns in air. When paper	
burns, the energy escapes as the heat and light of the flames. In a cell, the reaction occurs i	in
a more controlled way, and some of the energy is captured for use by the cell.	
Energy released by the exergonic "burning" of glucose in cellular respiration is	
used to make a substance called 14 A molecule of 15	
and a ¹⁶ group are joined to form each molecule of ATP. This is an en	d-
ergonic reaction, because it takes energy to assemble ATP. The covalent bond connecting	
the phosphate group to the rest of the ATP molecule is unstable and easily broken. This	
arrangement of atoms stores ¹⁷ energy. The ¹⁸ of	
ATP is an exergonic reaction. When ATP undergoes hydrolysis, a ¹⁹	
is removed, ATP becomes ²⁰ , and energy is released. Thus, ATP is a	
kind of energy "currency" that can be used to perform cellular 21 Moreover, 21	ost
cellular activities depend on ATP energizing other molecules by transferring its phosphate	e
group to them—a process called 22 It should be noted that energy is	
not destroyed when ATP is used to do work. When an ATP molecule is hydrolyzed to ma	
muscles move, some of its energy moves the body, and some ends up as random molecula	ar
motion, or ²³	
A less obvious but important function of ATP is supplying the energy for fighting	
in natural tendency for a system to become disordered. A cell constantly needs to manufa	C~
ture molecules to replace ones that are used up or damaged. Building a large molecule fror	n
$\frac{1}{2}$ muller parts is an 24 reaction. Energy released by the exergonic hydro	-
yell of ATP is used to drive essential endergonic reactions. The linking of exergonic and	
25 and ATP is the critical connection 25	<u>-</u>
Hun between the processes that release energy and those that consume it.	
What prevents a molecule of ATP from breaking down until its energy is needed?	•
Molecules can break down spontaneously; that is why ATP energy is needed to repair the	m.
Intuitively for living things, it takes some additional energy, called energy of 26	
that pre-	
Fills molecules from breaking down spontaneously. Energy barriers exist for both exer-	
multional endergonic reactions. Most of the time, most molecules in a cell lack the extra	
There's needed to clear the barrier, so chemical reactions occur slowly, if at all.	

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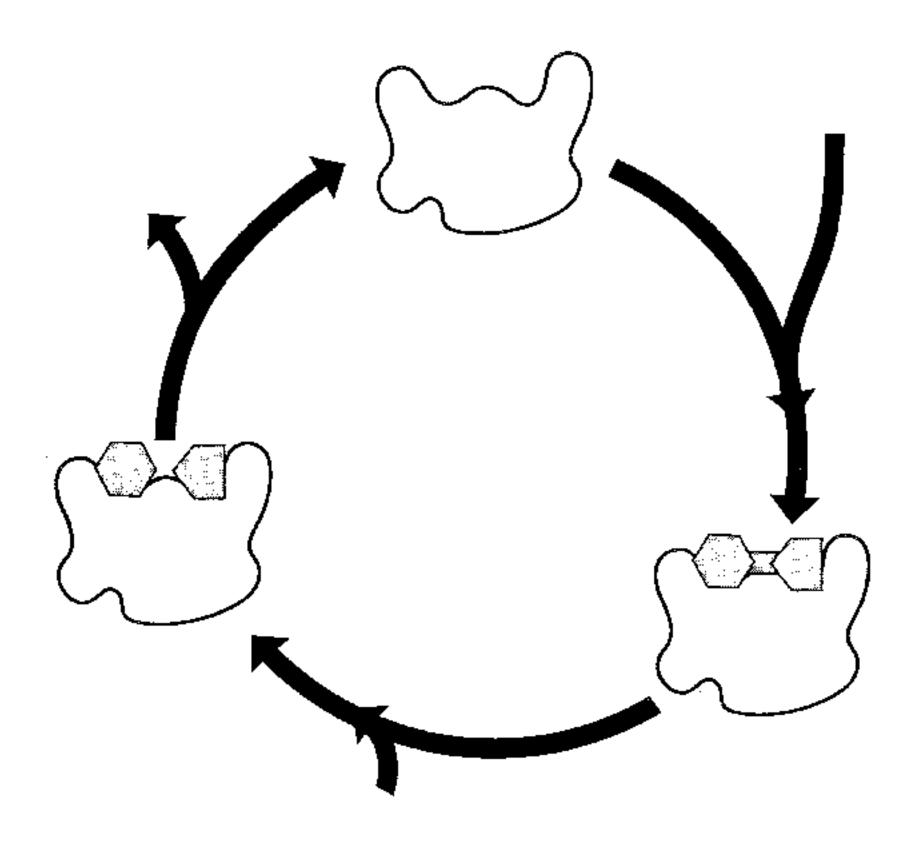
So what e	enables the vital reactions of more 1 - 1.	
needed, at a rate s	enables the vital reactions of metabolism to occur when and where the sufficient to sustain life? This is where enzymes come in. An enzyme is	v are
special ²⁸	molecule that are substantials where enzymes come in. An enzyme is) a
JU	the rate of a slave of the rate of a slave of the rate of a slave of the rate	
An enzyme holds	the rate of a chemical reaction without being 31 reactants in such a way as to 32 the energy barrier from reacting. Even though reactants would be in the energy barrier	ny it
that prevents then	o from reaction E	· y 11.
activation energy.	n from reacting. Even though reactants would not normally possess the needed to start the reaction, the enzyme grants.	_
reaction possible 1	needed to start the reaction, the enzyme creates conditions that make t	e 1
Where they are	Enzymes enable the cell to carry out vital chemical changes when and eded, enabling it to control the many chemical changes when and	ne
lular 33	eded, enabling it to control the many chemical reactions that make up of	_
		:el-
Example		
Exercise 2 (Mo	odules 5.1 – 5.5)	_
Web/CD Activity 5A	Fneray Transfer	I
Web/CD Activity 5B	57 ************************************	
Web/CD Activity 5C	Chemical Reactions and ATP The Structure of ATP	· "
	THE STRUCTURE OF ATP	
Briefly summarize t	the differences between the	
pairs.	the differences between the words or phrases in each of the following	
1. Kinetic energy a		
a diction chergy a	nd potential energy	
2 Evans.		
2. Exergonic reaction	ons and endergonic reactions	
2 D		
3. Reactants and pro	oducts	
	•	
4. ATP and ADP		
5. A reaction without	an enzyme and a reaction with an enzyme	
	an enzyme	
	•	
6. Photosynthesis and	cellular respiration	
7. First and second law	vs of thermodynamics	
	or dieimodynamics	

Exercise 3 (Modules 5.6 - 5.9)

Web/CD Activity 5D How Enzymes Work

Review enzyme action by completing the activities below.

1. Complete the diagram below so that it shows the cycle of enzyme activity. Imagine that the reaction carried out by this enzyme is splitting a substrate molecule into two parts. Color the diagram as suggested, and label the items in **boldface** type. Color the **enzyme** purple. Sketch the **substrate** as a dark pink shape. Sketch the **products**, and color them light pink. Also label the **active site**.



Make a sketch showing how heat or change in pH might change the above enzyme and alter its ability to catalyze its chemical reaction. Color and label the enzyme, its active site, and its substrate, as above.

On the left side of the space below, make a sketch showing how a competitive inhibitor might interfere with activity of the enzyme. Label the **competitive inhibitor**, and color it orange. On the right side, make a sketch showing how a noncompetitive inhibitor might interfere with activity of the enzyme. Label the **noncompetitive inhibitor**, and color it red.

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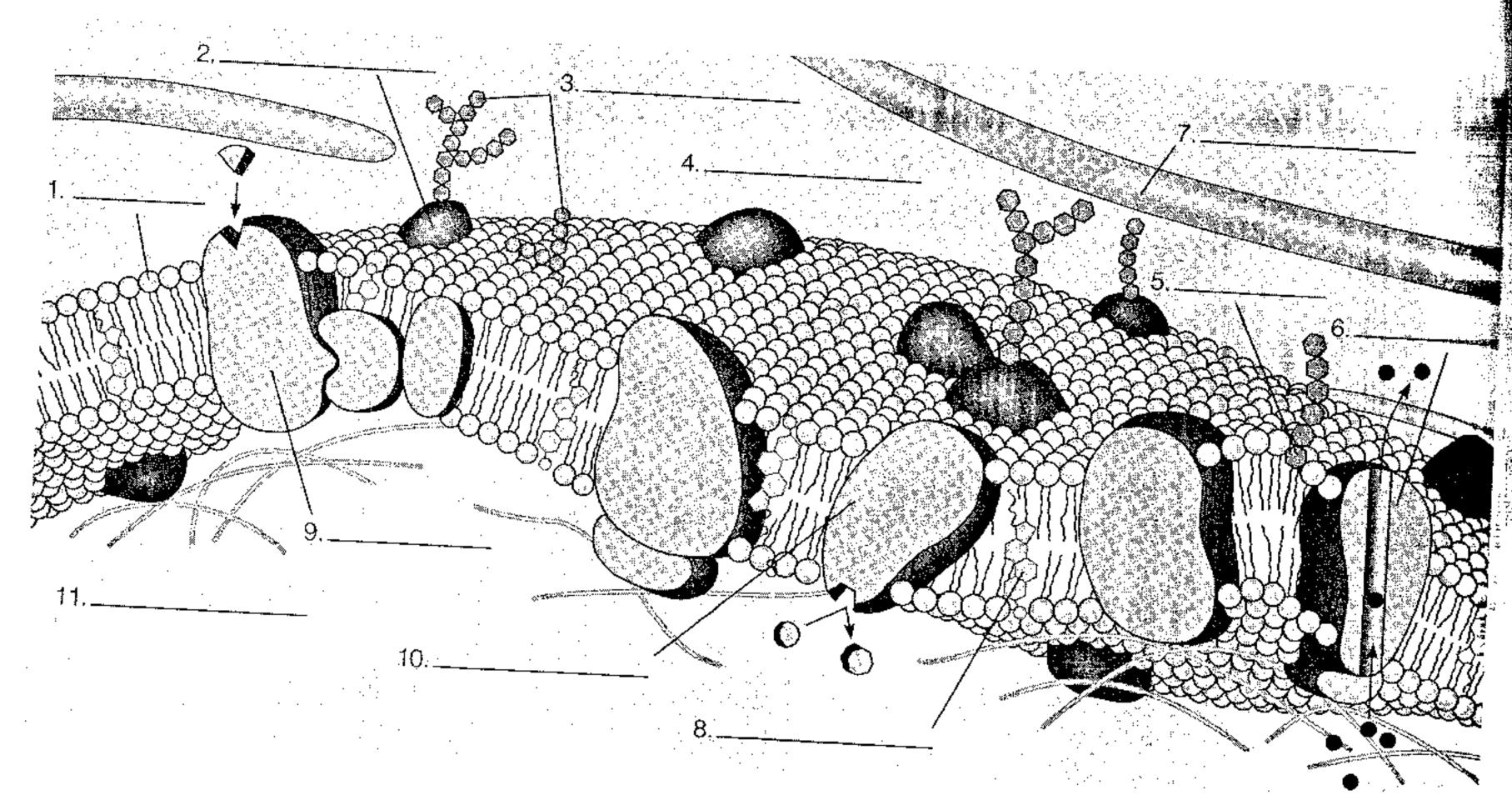
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Exercise 4 (Modules 5.10 - 5.13)

Web/CD Activity 5E Membrane Structure
Web/CD Activity 5F Signal Transduction

Web/CD Activity 5G Selective Permeability of Membranes

Review fluid mosaic membrane structure by coloring and labeling this diagram. It is a composite based on the figures in Modules 5.12 and 5.13. Label the items in **boldface** type: Start with the **cytoplasm**, **extracellular fluid**, and a **fiber of the extra cellular matrix**. In the membrane, color **phospholipids** gray, protein molecules purple, **carbohydrate I.D. tags** on **glycoprotein** and **glycolipid** molecules green, and **cholesterol** molecules yellow. Also show the functions of certain proteins by labeling them **enzyme**, **receptor protein**, and **transport protein**.



Exercise 5 (Modules 5.14 - 5.20)

Web/CD Activity 5H Diffusion

Web/CD Activity 51 Osmosis and Water Balance in Cells

Web/CD Activity 5J Facilitated Diffusion
Web/CD Activity 5K Active Transport

Web/CD Activity 5L Exocytosis and Endocytosis

Review diffusion and the function of cell membranes by matching each of the phrases on the right with the appropriate mechanisms from the list on the left. Two questions require more than one answer.

G. Pinocyt	transport is 2. Moves solutes against concentration gradient 3. Any spread of molecules from area of higher concentration to area of lower concentration 4. Diffusion with help of transport protein 5. Three types of endocytosis 6. Engulfing of fluid in membrane vesicles 7. Diffusion of water across selectively permeable mem-			
Evereise 6	(Modules 5.15 – 5.16)			
Web/CD Activ	vity 5l Osmosis and Water Balance in Cells			
the cell (In)	n important process that has many effects on living things. Test your underosmosis by predicting in each of the following cases whether water will enter or leave the cell (Out), or whether there will be no net movement of water me that the plasma membrane is permeable to water but not solutes.			
	Cell is exposed to hypertonic solution.			
2.	Cell is placed in salt solution whose concentration is greater than cell contents.			
3. Due to disease, solute concentration of body fluid outside cell is less than solute concentration of cells.				
4. (Cell is in isotonic solution.			
5. 9	Single-celled organism is placed in drop of pure water for examination under microscope.			
(our is mulicised in solution of sucrose and glucose whose individual concentrations			
ÿ <u></u>	cos than concentration of solutes in cytoplasm, but whose combined concentration is greater			
•	ent concentration of solutes in cytoplasm.			
7. 9	Solute concentration of cell is greater than solute concentration of surrounding fluid.			
· · · · ·	Len is exposed to hypotonic solution.			
9. C	Concentration of solutes in cytoplasm is equal to solute concentration of extracellular fluid. Cytoplasm more dilute than surrounding solution.			

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Exercise 7 (Modules 5.10 - 5.20)

Web/CD Activity 5E Membrane Structure Web/CD Activity 5F Signal Transduction

Web/CD Activity 5G Selective Permeability of Membranes Web/CD Activity 5H

Diffusion

Web/CD Activity 51 Osmosis and Water Balance in Cells

Web/CD Activity 5J Facilitated Diffusion Web/CD Activity 5K Active Transport

Web/CD Activity 5L Exocytosis and Endocytosis

Try to picture membranes and their functions close up by completing the following story.

Your first mission as a Bionaut requires you to enter a blood vessel and observe the structure and functions of cell membranes. You step into the water-filled chamber of the Microtron, which quickly shrinks you to a size much smaller than a red blood cell.

You tumble through the tunnel-like needle and into a blood vessel in the arm of a volunteer. Huge, rubbery red blood cells slowly glide past. Floating in the clear, yellowish blood plasma, you switch on your headlamp and examine the epithelial cells of the vessel wall. Their plasma membranes seem made of millions of small balloons. These are the polar "heads" of the ¹_____ molecules that make up most of the membrane surface. Through the transparent surface, you can see their flexible, tails projecting inward toward the interior of the membrane, and beyond them an inner layer of ³_____ molecules with their tails pointing toward you. Here and there there are globular ⁴_____molecules embedded in the membrane; some rest lightly on the surface, but most project all the way into the interior of the cell. The membrane is indeed a ⁵_____ mosaic; the proteins are embedded like the pieces of a picture, but you can see that they are free to move around. You push on one of the proteins, and it bobs like an iceberg. Some of the phospholipids and proteins have chains of sugar molecules attached to them, forming and ⁷_____. These are the molecules that act as cell tags. You notice that one of the proteins has a dimple in its surface. Just then a small, round molecule floating in the plasma nestles in the dimple. The molecule is a hormone, a chemical signal, and the dimpled protein is the ⁹______ In your light beam, you can see the sparkle and shimmer of many molecules, large and small, in the blood and passing through the cell membrane. Oxygen is moving from the plasma, where it is more concentrated, to the cell interior, where it is less concentrated. This movement is ¹⁰_____; when it occurs through a biological membrane, it is called ¹¹______transport. Similarly, carbon dioxide is

concentrated. You note that water molecules are passing through the membrane equally in both directions. The total concentration of solutes in the cell and in the blood must be

where it is ¹³_____ concentrated, to the blood, where it is

flowing out of the cell, down its ¹² gradient, from the cell interior,

equal; the solutions must be ¹⁵	You sig	mal the control team to inject a
small amount of concentrated salt soluti	on into the blood, r	naking the blood slightly
relative to the cell	l contents. This caus	ses water to flow
7 17		gain in equilibrium. This dif-
fusion of water through a ¹⁸	permeable	membrane is called
Some sugar molecules floating in through gaps in the membrane like the molecules simply bounce off, unless they proteins. This is a move down a concentration gradient with Because transport proteins help out, it is	nuch smaller water y happen to pass the type of passive tran thout the expenditu	molecules can. The sugar rough pores in special asport, because the molecules
Your chemscanner detects a high Transport proteins here and there in the cell against the concentration gradient. The the cell expends ²⁴ to the cell.	membrane are able his must be ²³	to move potassium into the transport:
if the substance is a solid particle. Sudder	rell the size of a build list of its membrane and general the body from a factor of the body from a factor of the pressure dingless.	lding quickly pins you are pressed against your face foreign invader! Taking in a specifically ²⁶ ninishes, and you are inside
the white blood cell, floating free in a mer	mbrane-enclosed ba	ag, or ²⁷
Another sac is approaching; it is a ²⁸ nanage to get your legs outside of the vac of the cell membrane. As the vacuole fuse and swim away from the voracious cell, reyou almost as fast as endocytosis trapped	, ful cuole and move it b s with the membra ealizing that ²⁹	l of digestive enzymes. You back toward the inner surface ne, you tear your feet free
You swim to the exit point, and the quite enough adventure for one day.	ne control team rem	oves you by syringe. This is